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## COMMUNICATION

# First total synthesis of antrocamphin A and its analogs as anti-inflammatory and anti-platelet aggregation agents<sup>†</sup>

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Naturally occurring antrocamphin A (1) is a potent antiinflammatory compound from the edible fungus *Antrodia camphorata* (*Taiwanofungus camphoratus*), whose wild fruiting body is used as a valuable folk medicine in Taiwan. This study is the first total synthesis of antrocamphin A (1) and its analogs. Their inhibition ability on NO release, superoxide anion generation, elastase release and platelet aggregation are reported herein.

#### Introduction

The endemic fungus *Antrodia camphorata* (*Taiwanofungus camphoratus*), also known as Niu-Chang-Chih (Jang-Jy), is used as a folk medicine and a dietary supplement in Taiwan. Its wild fruiting body is very valuable.<sup>1-3</sup> The chemical composition of this fungus can be classified into three categories: 1). Polysaccharides, 2). Triterpenoids and 3). Enynyl-benzenoids. The polysaccharides are regarded as immunomodulation agents, such as functional foods derived from other fungi, *e.g., Ganoderma lucidum* and *Agaricus blazei*.<sup>1</sup> Triterpenoids are considered to have anti-cancer and anti-inflammatory effects. However, the major chemical component, antrocamphin A (1) (Fig. 1), which belongs to the third class, is the key component for anti-inflammatory activity. Evidence is

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Fig. 1 Antrocamphin A (1).

shown below through a complete bioassay-guided fractionation study.  $^{\rm 3.4}$ 

Antrocamphin A (1), a potent anti-inflammatory compound naturally found in *A. camphorata* (*T. camphoratus*) and was first isolated by Chen *et al.* in 2007.<sup>4</sup> Wang *et al.* demonstrated its mechanism in suppressing pro-inflammatory molecules (NO and PGE<sub>2</sub>), from being released *via* the down-regulation of iNOS and COX-2 expression through the NF- $\kappa$ B pathway.<sup>3</sup> Previous studies indicate that compound 1 may serve as a promising lead drug for the treatment of various diseases that are induced by inflammation. Therefore, antrocamphin A was chosen to be a candidate of further investigation.

Currently, there are only two literature<sup>3,4</sup> reports on the subject of antrocamphin A (1). To understand the diverse biological properties and the structure–activity relationship (SAR) of the lead compound 1, the first total synthesis of compound 1 and its analogs were achieved in the current investigation. All compounds were evaluated for anti-inflammatory and anti-platelet aggregation activities.

#### **Results and discussion**

The retrosynthetic analysis of compound 1 is illustrated in Scheme 1. The key step was the Sonogashira reaction. Compound 2 (2-iodo-3,5,6-trimethoxytoluene) was coupled with 2-methyl-1buten-3-yne (3), which led to the target compound. Compound 2 was synthesized from 2,3,5-trimethoxytoluene (8), obtained from *o*-vanillin (5), in four steps described by Singh and co-workers.<sup>5</sup> We prepared 2-hydroxy-3-methoxytoluene (4) by hydrogenating of *o*-vanillin (5) over 10% palladium-on-carbon. Compound 4 was then converted to quinone 6 by treatment with the oxidant, potassium nitrosodisulfonate (Fremy's salt). Quinone 6 was reduced to hydroquinone 7 using TiCl<sub>3</sub> and then methylated with Me<sub>2</sub>SO<sub>4</sub> in the presence of K<sub>2</sub>CO<sub>3</sub> resulting in the desired compound 8. Compound (2), 2-iodo-3,5,6-trimethoxytoluene, a key synthon,

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Scheme 1 Retrosynthetic analysis of antrocamphin A.

was prepared *via* iodination of **8** using  $I_2$  and CF<sub>3</sub>COOAg. The Sonogashira reaction was utilized to couple iodo **2** with 2-methyl-1-buten-3-yne (**3**), using the catalysts Pd(PPh<sub>3</sub>)<sub>4</sub> and CuI. This key reaction led to the desired compound, antrocamphin A (**1**) (Scheme 2).<sup>6</sup> The Sonogashira coupling reaction was also applied to afford a series of designed analogues (**9–22**) from commercially available iodobenzene products or further iodination benzenoid compounds (Scheme 3).

Synthesized compounds **1** and **9–22** were screened using a nitric oxide (NO) inhibitory assay, where curcumin was used as the positive control. In this study, all analogs were tested on their cytotoxicity toward RAW 264.7 macrophages. Cell viability rates for each test compound showed > 90% at the dosage of 20  $\mu$ g mL<sup>-1</sup>. All compounds were evaluated against lipopolysaccharide (LPS)-induced NO production in RAW 264.7 macrophage cell line at the dosage of 20  $\mu$ g mL<sup>-1</sup>. As shown in Table 1, compounds **1**, **20**, **21** and **22** showed potent anti-inflammatory activity with NO inhibition rates of 98.1%, 97%, 77%, 84%, respectively.

Compounds 1, 9, 15, 17, 20 and 21 were synthesized to study the SAR of the  $CH_3$  substitution at C-3 presented in the natural product 1; however, C3-CH<sub>3</sub> did not influence the ability of NO inhibition. To evaluate the SAR of the OCH<sub>3</sub> functions on the 3'methyl-but-3'-en-1-ynyl-benzene skeleton, analogs 10–12, 15 and



Compound -	Substituent group				Yield
	R1	$R_2$	R <sub>3</sub>	$R_4$	(%)
9	Η	Н	Н	Н	93
10	Η	Н	$OCH_3$	Н	87
11	OCH <sub>3</sub>	Н	Н	Н	73
12	Н	OCH <sub>3</sub>	Н	Н	55
13	Н	Н	$CH_3$	Н	89
14	Н	Н	CN	Н	83
15	Н	$OCH_3$	$OCH_3$	$OCH_3$	81
16	Η	Н	$NO_2$	Н	83
17	$CH_3$	Н	Н	Н	77
19	$CH_3$	OCH <sub>3</sub>	OCH <sub>3</sub>	Н	73
20	Н	Н	$OCH_3$	$OCH_3$	63
21	$CH_3$	Н	$OCH_3$	$OCH_3$	17
22	CH <sub>3</sub>	Ι	OCH <sub>3</sub>	OCH <sub>3</sub>	13

Scheme 3 Synthesis of 9–22. Reagents and conditions: (a) 2-methyl-1-buten-3-yne (3), Pd(PPh\_3)<sub>4</sub>, CuI, Et<sub>3</sub>N/THF, N<sub>2</sub>, rt.

**20** were designed, which did not have the C3-CH<sub>3</sub> substitution. Compounds **10–12** possess only one OCH<sub>3</sub> group at C-1, C-3 and C-2, respectively. Moreover, **20** and **15** have two and three OCH<sub>3</sub> groups at C-1 and C-5, and C-1, C-2 and C-5, respectively. Among the five compounds, we speculated that the position is more important than the number of OCH<sub>3</sub> substitutions. A similar result was also observed in the test of compounds **19** and **21**, which both have a CH<sub>3</sub> group at C-3 and the same number of OCH<sub>3</sub> substitutions in different positions. Furthermore, a comparison



Scheme 2 Synthetic scheme of antrocamphin A (1). Reagents and conditions: (a) 10% Pd/C, H<sub>2</sub>, rt, 75%; (b) KH<sub>2</sub>PO<sub>4</sub>, (KSO<sub>3</sub>)<sub>2</sub>NO, rt, 78%; (c) TiCl<sub>3</sub>, rt, 90%; (d) (CH<sub>3</sub>O)<sub>2</sub>SO<sub>2</sub>, K<sub>2</sub>CO<sub>3</sub>, acetone, reflux, 88%; (e) I<sub>2</sub>, CF<sub>3</sub>COOAg, CH<sub>2</sub>Cl<sub>2</sub>, 0 °C, 74%; (f) 2-methyl-1-buten-3-yne (3), Pd(PPh<sub>3</sub>)<sub>4</sub>, CuI, Et<sub>3</sub>N/THF, N<sub>2</sub>, rt, 10%.

 Table 1
 Effects of all compounds on NO in LPS-challenged RAW 264.7

 cells<sup>a</sup>

Compound	NO inhibition (%)	Cell viability (%)	
1	98.10	99.71	
9	17.23	101.55	
10	33.56	101.98	
11	-4.43	90.44	
12	27.27	105.80	
13	23.11	100.95	
14	-0.55	91.77	
15	24.98	101.91	
16	15.16	94.97	
17	6.23	99.73	
18	16.61	92.23	
19	28.51	99.07	
20	97.00	117.08	
21	77.00	114.00	
22	84.00	102.00	
Curcumin <sup>c</sup>	99.00	103.00	

<sup>*a*</sup> All compounds were administrated 1 h before inflammation induction by adding LPS (1  $\mu$ g mL<sup>-1</sup>). <sup>*b*</sup> Dosage of test compound was 20  $\mu$ g mL<sup>-1</sup>. <sup>*c*</sup> Positive control.

with the structures of 1 and 20–22 concluded that the simultaneous existence of OCH<sub>3</sub> groups at C-1 and C-5 may be necessary for the potential ability of NO inhibition.

LPS-challenged RAW 264.7 macrophages are an in vitro model of murine cell lines for studying anti-inflammatory effects. To further explore the anti-inflammatory activities of these target compounds in human cell systems with different modes of action, all analogs were evaluated against superoxide anion generation and elastase release by human neutrophils in response to N-formyl-methionyl-leucyl-phenylalanine (FMLP)/cytochalasin B (CB). Antrocamphin A has been reported as a new anti-inflammatory drug lead against superoxide anion generation in FMLP-activated human neutrophils.<sup>4</sup> Activated neutrophils produced high concentrations of the superoxide anion and elastase known to be involved in airway mucus hypersecretion, which is a neutrophilic inflammatory disease like asthma.<sup>7</sup> Therefore, all analogues were tested for their ability to inhibit superoxide anion generation and elastase release. Diphenyleneiodonium (DPI) and phenylmethylsulfonyl fluoride (PMSF) were used as positive controls in this model assay, respectively.

Interestingly, the analogs showed a broad-spectrum of activity in this model (Table 2). Most of the compounds could inhibit both inflammatory mediators and showed more potent effects than the natural product 1. Compound 18 with a dienynyl functionality exhibited the best activity, which indicated the enynyl substitution played an important role for this model assay. On the whole, we found dimethoxy (19–22) and monomethoxy (10–12) derivatives were more favorable than trimethoxy and non-methoxy ones. Some of the compounds showed superior activities to the reference drug, PMSF, against elastase release by neutrophils.

Furthermore, all synthesized derivatives were also subjected to an anti-platelet aggregation assay with collagen and thrombin as inducers; aspirin was used as a positive control. As shown in Table 3, compounds **9**, **10**, **11** and **19** showed potent inhibition against collagen-induced platelet aggregation with IC<sub>50</sub> values of 3.36, 1.16, 7.55 and 7.22  $\mu$ g mL<sup>-1</sup>, respectively; their activities were better than aspirin's activity (IC<sub>50</sub> 13.58  $\mu$ g mL<sup>-1</sup>). Comparing the

elastase release by human neutrophils in response to FMLP/CB Superoxide Elastase Compound  $IC_{50}/\mu g \, m L^{-1a}$ or (Inh %)  $IC_{50}/\mu g \, m L^{-1a}$ or (Inh %)  $(46.30 \pm 7.24)$ \*\*\*  $(25.55 \pm 3.99)$ \*\* 1 9 \*\*\*  $482 \pm 041$  $(37.22 \pm 3.12)$ 10  $5.63 \pm 0.71$  $4.14 \pm 0.35$  $6.69 \pm 0.84$  $7.94 \pm 0.51$ 11 12  $4.05\pm0.41$  $3.31\pm0.06$ 13  $4.66 \pm 0.41$  $44.28 \pm 1.67$ 14  $(7.95 \pm 6.06)$  $(17.89 \pm 1.42)$ 15  $3.40 \pm 1.49$  $(26.74 \pm 5.88)$ 16  $(-10.03 \pm 4.84)$  $22.78 \pm 1.01$  $(35.55 \pm 6.12)$ 17  $(25.55 \pm 6.70)$ \*\* 18  $0.45 \pm 0.03$  $1.32 \pm 0.67$ 19  $1.96 \pm 0.28$  $3.47 \pm 0.23$ 

 $8.72 \pm 1.15$ 

 $5.33 \pm 1.21$ 

 $(81.36 \pm 8.82)$ 

 Table 2
 Effects of all compounds on superoxide anion generation and

fluoride (PMSF) were used as positive controls for superoxide anion

 Table 3
 Anti-platelet aggregation data of all compounds

 $4.69 \pm 0.99$ 

 $1.60 \pm 0.28$ 

 $(12.13 \pm 2.24)$ 

generation and elastase release, respectively.

	$IC_{50}/\mu g \ m L^{-1}$			
Compound	Collagen (10 µg mL <sup>-1</sup> )	Thrombin (0.05 U/mL)		
1	>50	>50		
9	3.36	40.83		
10	1.16	44.06		
11	7.55	>50		
12	>50	>50		
13	>50	>50		
14	12.14	36.09		
15	15.43	>50		
16	>50	>50		
17	>50	>50		
18	14.50	12.92		
19	7.22	>50		
20	>50	>50		
21	>50	>50		
22	>50	>50		
Aspirin <sup>a</sup>	13.58	>200		
" Positive control				

bioassay results, most of the compounds did not possess good inhibition toward the platelet aggregation induced by thrombin. No clear SAR can be concluded from the anti-platelet aggregation effect. Interestingly, compounds 1 and 20–22, which possess the powerful ability of NO inhibition, did not have anti-platelet aggregation activity.

#### Conclusions

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In summary, we have achieved the first total synthesis of the active natural product antrocamphin A (1), in a total yield

of 3.7% in six steps. Moreover, 14 analogs, including new compounds, **15**, **16** and **18–22**, were synthesized. Most of these compounds possess OMe groups in the benzene ring. We tried to synthesized derivatives with different substitutions around the ring, for example, 2-iodo-3,6-diacetoxy-5-methoxytoluene or 4-iodo-3,6-dihydroxy-5-methoxytoluene coupled with 2-methyl-1-buten-3-yne using different combinations of catalyst [Pd(PPh<sub>3</sub>)<sub>4</sub> or PdCl<sub>2</sub>(PPh<sub>3</sub>)<sub>2</sub>], solvent (THF or DMF or ether) and base (Et<sub>3</sub>N). Unfortunately, this resulted in no reaction or reaction without any target compound. Accordingly, only the reagents with OMe groups, which were easily obtained, were chosen to evaluate the amount and position of OMe groups related to the activity in this investigation.

Comparing anti-inflammatory activity, compounds 1, 20, 21 and 22 were identified to be potent NO inhibition agents; the dienynyl compound 18 was a new drug lead against superoxide anion generation and elastase release. Moreover, analog 10 is the most potential compound against platelet aggregation induced by collagen. Overall, our data demonstrate that enynyl-benzenoids may have a chance to be developed as safe and potential anti-inflammatory or a cardiovascular protecting agent in the future.

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